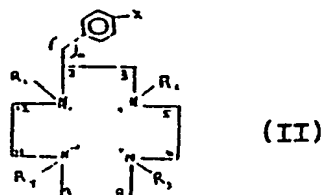
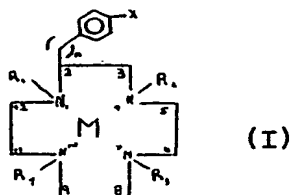


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| (51) International Patent Classification ⁴ : C07D 257/00, C07F 9/94 C07F 7/24, 5/06, 3/08 C07F 3/10, 5/00, 3/12 C07F 3/00 | A1 | (11) International Publication Number: WO 89/11475 (43) International Publication Date: 30 November 1989 (30.11.89) |
| (21) International Application Number: PCT/US89/02201 (22) International Filing Date: 24 May 1989 (24.05.89) (30) Priority data: 198,538 25 May 1988 (25.05.88) US (71) Applicant: THE UNITED STATES OF AMERICA, as represented by THE SECRETARY, UNITED STATES DEPARTMENT OF COMMERCE [US/US]; 5285 Port Royal Road, Springfield, VA 22161 (US). (72) Inventors: GANSOW, Otto, A. ; 3003 Van Ness, N.W., Washington, DC 20008 (US). BRECHBIEL, Martin, W. ; 3404 Monarch Lane, Annandale, VA 22003 (US). MARGERSTADT, Michael, A. ; Rheingaustrasse 19, D-6238 Hofheim (DE). | | (74) Agents: STERN, Marvin, R. et al.; Fleit, Jacobson, Cohn, Price, Holman & Stern, The Jenifer Building, 400 Seventh Street, N.W., Washington, DC 20004 (US). (81) Designated States: AT (European patent), AU, BE (European patent), CH (European patent), DE (European patent), FR (European patent), GB (European patent), IT (European patent), JP, LU (European patent), NL (European patent), SE (European patent). Published <i>With international search report.</i> |

(54) Title: MACROCYCLIC CHELATES AND METHODS OF USE THEREOF**(57) Abstract**

The invention is a chelate having general formula (I), wherein R_{1-4} is $-\text{CH}_2\text{COOH}$; n is 1 to 5; X is a member selected from the group consisting of $-\text{NO}_2$, $-\text{NH}_2$, $-\text{NCS}$, $-\text{NHCOCH}_2\text{-Z}$ with Z being a member selected from the group consisting of Br and I , $-\text{COOH}$; $-\text{OCH}_2\text{OOCH}$; and M is a metal ion being a member selected from the group of elements consisting of Bi , Pb , Y , Cd , Hg , Al , Th , Sr , and Lanthanides. The invention can include a chelate wherein M is a copper ion and n is an integer from 2 to 5. The invention includes chelate conjugates of general formula (I) and ligand conjugates of general formula (II). The invention also includes methods to use these compounds for treatment of cellular disorders and for diagnostic tests.

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1 MACROCYCLIC CHELATES AND METHODS OF USE THEREOF

2 BACKGROUND OF THE INVENTION

3 1. Field of the Invention

4 This invention relates to macrocyclic chelates and
5 methods of use thereof. More specifically, this
6 invention relates to 2-substituted 1, 4, 7, 10-Tetraaza
7 cyclododecane -N, N', N'' N'''-tetraacetic acid, and
8 2-substituted 1, 4, 7, 10-Tetraazacyclododecane, and
9 analog macrocycles and their uses.

10 2. Description of the Background Art

11 Macrocycles have been studied for their usefulness as
12 chelates for numerous metal ions that have therapeutic
13 diagnostic, or other uses. A macrocycle of particular
14 usefulness as a chelate is the 1, 4, 7, 10-Tetra-
15 azacyclododecane -N, N, N, N-tetraacetic acid (DOTA).
16 DOTA compounds have been linked to biomolecules to form
17 delivery systems for the chelated metal ion to specific
18 sites within an organism.

1 U.S. Patent Number 4,678,667 to Meares et al.,
2 discloses a macrocyclic bifunctional chelating agent
3 The chelating agents of this disclosure can include DOTA
4 compound that is a Cu(II) chelate. The usefulness of the
5 chelating agent is limited to the effects of the copper
6 metal ion. The synthesis of this disclosure gives low
7 and not always reproducible results.

8 An earlier U.S. Patent No. U.S. 4,622,420 to Meares
9 et al. disclosed bifunctional chelating agents of the
10 acyclic ligand ethylene diamene N, N' N'', N'''
11 tetraacetic acid (EDTA) useful for binding metals other
12 than copper such as Indium. These compounds are useful
13 for imaging of tumors.

14 U.S. Patent Number 4,652,519 to Warshawsky et al.,
15 discloses bifunctional chelating agents and process for
16 their production. The compounds disclosed in this patent
17 are analogues of EDTA. These compounds are used to
18 chelate metal ions and are linked to haptens to provide
19 specific site selection within an organism. The
20 compounds of this patent are offered to provide an
21 improved substituent for the EDTA compounds such as
22 those disclosed in the Meares et al. patent discussed
23 above.

24 U.S. Patent Numbers 4,454,106 and 4,472,509 to Gansow
25 et al., respectively disclose the use of metal chelate
26 conjugated monoclonal antibodies and the specific metal
27 chelate conjugated monoclonal antibodies. These
28 disclosures provide compounds and methods for treating
29 cellular disorders. Radiometal chelate conjugated
30 monoclonal antibodies specific to a target cell are used
31 to deliver alpha, beta, or Auger electron emitting metal
32 ions. These disclosures are not related to DOTA
33 compounds.

1 The value of having a ligand conjugate to chelate
2 metal ions for therapeutic, diagnostic, or other uses is
3 of commercial importance. This commercial importance is
4 created by the fact that many metal ions have desirable
5 characteristics for these various uses, but the delivery
6 systems for the metal ions lack specificity to target
7 cells or do not adequately bind the metal ions. Examples
8 of the usefulness of specific metal ions are as follows.

9 The usefulness of radionuclide materials in cancer
10 therapy is disclosed in the article, Kozak et al.,
11 "Radionuclide-conjugated monoclonal antibodies: A
12 Synthesis of Immunology, in Organic Chemistry and Nuclear
13 Science" Trends in Biotechnology. 4(10):259-264 (1985).
14 This article discusses the use of antibody conjugates to
15 deliver either alpha or beta radiation. The value of
16 alpha radiation from bismuth-212 in radionuclide therapy
17 is further discussed in the two articles, Kozak et al.,
18 "Bismuth-212-labeled anti-Tac monoclonal antibody:
19 Alpha-particle-emitting Radionuclides as Modalities for
20 Radioimmunotherapy" Proc. Natl. Acad. Sci. U.S.A.
21 83:474-478 (1986) and Gansow et al., "Generator-produced
22 Bi-212 Chelated to Chemically Modified Monoclonal
23 Antibody for Use in Radiotherapy" Am. Chem. So.
24 Symposium Series 15:215-227 (1984).

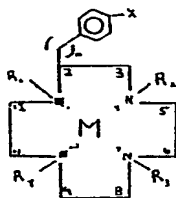
25 Examples of other uses for chelated metal ions are
26 disclosed in the following articles. Magerstadt et al.,
27 "Gd(DOTA): An Alternative to Gd(DPTA) as a T_{1,2}
28 Relaxation Agent for NMR Imaging or Spectroscopy"
29 Magnetic Resonance in Medicine 3:808-812 (1986),
30 discloses the usefulness of gadolinium as a relaxation
31 agent for NMR imaging. The article, Spirlet et al.,
32 "Structural Characterization of a Terbium(III) Complex
33 with 1, 4, 8, 11-Tetraazacyclotetradecane- 1, 4, 8,

1 11-tetraacetic Acid. Lanthanide Ions and the
2 Conformation of the 14-Membered Macrocyces" Inorganic
3 Chemistry 23(25):4278-4283 (1984), discloses the
4 usefulness of lanthanide chelates.

5 The industry is lacking a DOTA chelate that can be
6 efficiently produced in high yields and that has
7 desirable chelating qualities for numerous metal ions.

8 SUMMARY OF THE INVENTION

9 The invention is a chelate having a general formula
10 I:



11 wherein R_{1-4} is $-\text{CH}_2\text{COOH}$;

12 n is 1 to 5;

13 X is a member selected from the group consisting of

14 $-\text{NO}_2$,

15 $-\text{NH}_2$,

16 $-\text{NCS}$,

17 $-\text{NHCOCH}_2 - \text{Z}$ with Z being a member selected from the
18 group consisting of Br and I

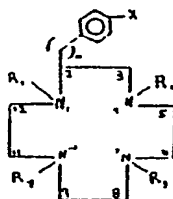
19 $-\text{COOH}$;

20 $-\text{OCH}_2\text{COOH}$

21 and M is a metal ion being a member selected from the
22 group of elements consisting of

23 Bi , Pb , Y , Cd , Hg , Al , Th , Sr , and Lanthanides.

The invention can include a chelate wherein M is a copper ion and n is an integer from 2 to 5. The invention includes chelate conjugates of general formula I and ligand conjugates of general formula II:



The invention also includes methods to use these compounds for treatment of cellular disorders and for diagnostic tests.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 illustrates a chemical pathway to produce the preferred embodiment of the invention.

DETAILED DESCRIPTION OF THE INVENTION

The compound of this invention is a substituted DOTA represented by the general formula I shown above or specifically by compound X of Figure 1. Compound X can subsequently be converted to other substituted DOTA compounds, but compound X is the parent compound for such other compounds. The general formula is a 12 membered ring tetraaza macromolecule with the nitrogens in the 1, 4, 7, and 10 positions. Each of the nitrogens is "ribbed" by an ethylene group.

The substituted DOTA ligand represented by compound X of Figure 1 complexes metals. Metal complexes are formed by placing the DOTA into solution with an appropriate metal salt having the metal to be chelated. Metal salts

1 have to be selected so as to prevent the hydrolysis of
2 the metal. Also, reaction conditions in an aqueous medium
3 have to be chosen such that the metal is not hydrolyzed.
4 For example, a lead nitrate complex, bismuth iodide
5 complex, or yttrium acetate salts can be used to form a
6 metal chelate with lead, bismuth, or yttrium,
7 respectively. General examples of suitable salts include
8 any soluble divalent metal complex or any trivalent metal
9 complex that is not hydrolyzed at pH 4 or below. Thorium
10 requires the use of iodide salt, specifically. The most
11 desirable metal ions for chelation with general formula 1
12 are members from the group consisting of bismuth, lead,
13 yttrium, cadmium, mercury, actinium, thorium, strontium,
14 and any of the elements of the lanthanide elements. The
15 most desirable elements of the lanthanide series are
16 gadolinium, for use in NMR imaging and as a relaxation
17 agent in NMR imaging, and terbium and europium because of
18 their use as chromophores in time resolved fluorescence
19 spectroscopy. These fluorescent compounds can be useful
20 in an in vitro diagnostic assay where a fluorescent assay
21 is used rather than a radioactive amino assay.

22 The X substituent of general formula is desirably a
23 substituent that conjugates the compound with haptans.
24 This substituent is desirably a free-end nitro group
25 which can be reduced to an amine. The amine can then be
26 activated with a compound such as thionyl chloride to
27 form a reactive chemical group such as an
28 isothiocyanate. An isothiocyanate is preferred because
29 it links directly to amino residues of a haptan such as a
30 monoclonal antibody. The amine group can be linked to
31 an oxidized carbohydrate on the protein and,
32 subsequently, the linkage is fixed by reduction with
33 cyanoborohydride. The amino group can then also be
34 reacted with bromoacetyl chloride or iodoacetyl chloride

1 to form $\text{-NHCOCH}_2\text{Z}$ with Z being bromide or iodide. This
2 group reacts with any available amine or sulfhydryl group
3 on a hapten to form a stable covalent bond. If tyrosine
4 is used in the formulation of the macromolecule a
5 carboxylic acid or methoxy carboxylate group can be in
6 this position of the compound. The most desirable
7 substituents for this position are members selected from
8 the group consisting of -NO_2 , -NH_2 , -NCS , -COOH ,
9 $\text{-OCH}_2\text{COOH}$, $\text{-OCH}_2\text{COOH}$ and $\text{-NHCOCH}_2\text{-Z}$ with Z being a member
10 selected from the group consisting of bromide and
11 iodide. The preferred substituent for this position is
12 -NCS .

13 The haptens suitable for linking with the substituent
14 at the X position of general formula I can vary widely.
15 The most desirable haptens are members selected from the
16 group consisting of hormones, steroids, enzymes, and
17 proteins. These haptens are desirable because of their
18 site specificity to tumors and/or various organs of the
19 body. The preferred hapten for use in treating cellular
20 disorders or various disease conditions is a monoclonal
21 antibody.

22 The compound of this invention can have n equal an
23 integer from 1 to 5. In the preferred embodiment, M
24 equals 2. It is desirable for n to equal 2 versus 1
25 because the chelating ligand is further separated from
26 the antibody and has more rotation. The increased free
27 rotation allows a metal to chelate with the macromolecule
28 more easily. When n is 3 or greater, the synthesis of
29 the compound becomes lengthy.

30 Figure 1 illustrates the preferred reaction pathway
31 or process for forming the compound of this invention.
32 This reaction results in a compound of general formula I

wherein n is 1. If n is to equal 2, an additional methylene group would be present between the alpha amino carbon and the aromatic group. This compound is 2-amino-4-nitrophenylbutyric acid.

The process for synthesizing a compound according to this invention first provides a triamine with a substituent in the 2-position. The embodiment of Figure 1 has a methylene [n = 1] as the initial substituent for linkage. The preferred embodiment has a phenylethylenylene group. The process then provides a tetraaza macromolecule having the substituent in the 2 position. Alkylation with bromoacetic acid to form the four carbon to nitrogen bonds of the carboxymethylene substituents at the R1, R2, R3, and R4 in the general formula.

The process of Figure 1 reacts p-nitrophenyl alanine with methanol and hydrochloric acid to form the ester compound II. This ester is reacted with ethylenediamine in the presence of triethylamine to remove the hydrochloride salt of the ester formed in compound II. The condensate of the amide of the ethylenediamine adduct or compound III is subsequently reacted with a diactive ester or compound VI to form a cyclic product or compound VII.

The desired diactive ester VI is formed sequentially from amidodiacetic acid for IV of Figure 1. The amine is first blocked by using the reagent BOC-ON or any other suitable blocking agent, such as FMOC, in the presence of triethylamine which serves to deprotonate the starting material. The subsequent nitrogen blocked diacetic acid V or other such nitrogen blocked compound is then coupled to N-hydroxysuccinimide, or any other suitable compound such as phenols, or N-hydroxydicarboximides which forms a

1 reactive ester. The choice of compounds which form
2 active esters or blocking groups is within the scope of
3 the art. The coupling is done by dicyclohexyl-
4 carbodiimide or "DCC". This step produces the nitrogen
5 blocked active ester or compound VI.

6 Ring formation under high dilution conditions between
7 amino acidamide or compound III with the nitrogen blocked
8 active ester of compound VI then occurs. This condensing
9 step forms the triamide macrocycle or compound VII.
10 Compound VII is produced in very high yield. The yield
11 is typically at least about 80 percent. The yield more
12 desirably is between about 80 percent to about 95
13 percent.

14 The synthesis of the macrocycle of compound VII may
15 be accomplished by two pathways. The amine nitrogen of
16 compound VII is deblocked with trifluoroacetic acid or
17 "TFA". This forms the TFA salt of the triamide
18 macrocycle or compound VIII. This compound is reduced
19 with borane/petrahydrofuran or THF. The resulting borane
20 adduct is cleaved by hydrochloric acid to form the
21 substituted tetraazamacrocycle of compound IX. This
22 tetraazamacrocycle can then be alkylated with haloacetic
23 acid in the presence of base to form a nitrobenzyl DOTA
24 or compound X. Alternatively, compound VII can be
25 reduced with borane/THF and reacted with hydrochloric
26 acid to form compound IX directly. This alternative
27 pathway produces slightly poorer yields.

28 The nitro group of compound X can be reduced with
29 hydrogen over platinum on a carbon catalyst to produce
30 the amino group or the aminobenzyl DOTA depicted as

1 compound XI. Compound XI can then be reacted with
2 thiophosgene to produce the isothiocyanate or compound
3 XII.

4 The methodology in column 3 of U.S. Patent Number
5 4,652,519 to Warshawsky et al., hereby incorporated by
6 reference, provides the methodology to produce the -COOH
7 substituent. This procedure produces the ethylene
8 diamine intermediate. The desired intermediate
9 macrocycle is produced by forming the analogous diactive
10 ester of compound VI by using N,
11 N'-ethylenediaminediacetic acid. Condensation of the
12 diamine with the dinitrogen, diBOC diactive ester
13 produces diamide intermediary which is reduced by
14 diborane to produce the appropriate tetraaza macrocycle.
15 The DOTA ligand can be made from this macrocycle. The
16 synthesis of the X and Z groups are also disclosed by the
17 Warshawsky patent.

18 The reaction steps described above to produce
19 compounds X, XI, and XII are known. The novel feature of
20 the process of Figure 1 is the cyclization procedure.
21 The conversion reaction of compound IV with compound VI
22 to form the macrocycle and the full reduction of the
23 macrocycle to produce compound X produces the unexpected
24 results of very high yields compound X.

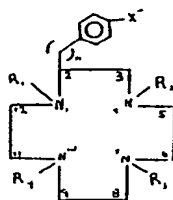
25 In its preferred embodiment the coupling of an
26 isothiocyanate chelate of compound XII of Figure 1 is
27 done by direct conjugation of the isothiocyanate with a
28 free amino group found in many residues of proteins,
29 enzymes or other compounds such as certain hormones. An
30 example of this situation with a hormone is found with
31 the free amino group provided by the epsilon amino group
32 of the lysine or the terminal amino group as the hormone

1 peptide chain. Any free amino group can react with the
2 isothiocyanate to form a thiourea linkage which is
3 covalently coupled and irreversible. The use of a
4 steroid as a hapten requires that an amino function be
5 present in the steroid.

6 An advantage of the amine derivative chelate of
7 compound XI of Figure 1 is that, when coupling to
8 proteins and, in particular, when coupling to antibodies,
9 the carbohydrate of the antibody can be oxidized prior to
10 the coupling reaction. The amine reacts with the
11 aldehyde that is formed on the protein. This aldimine
12 formed can be reduced by cyanoborohydride to form a
13 covalent secondary amine linkage to the antibody in a
14 position that is site specific. This position is away
15 from the binding site of the FAB'2 part of the monoclonal
16 antibody.

17 A desirable embodiment of the invention is one having
18 copper metal ion and n is an integer from 2 to 5. This
19 embodiment of the invention can be used to label a
20 monoclonal antibody with Cu^{67} . When n is an integer from
21 2 to 5, there is less hindrance of the chain of the
22 ligand with the protein than occurs when n is 1. When n
23 is an integer from 2 to 5, sufficient space is provided
24 between the ligand and the protein to allow freer
25 rotation of the ligand. This results in more efficient
26 chelation of the copper ion by the resulting conjugate.

27 An embodiment of the invention involves a
28 ligand-hapten conjugate of general formula II:



1 This conjugate chelates metal ions. It is desirable to
2 expose many metals to the protein conjugate in a
3 concentrated metal solution for as short a period of time
4 as possible. Certain metals, such as divalent metal
5 ions, react rapidly and directly with the conjugate. The
6 kinetics of the formation reactions for these compounds
7 are so rapid that it is desirable to have the
8 ligand-hapten conjugate available in the pharmacy
9 immediately prior to use. The conjugate can then be
10 mixed in the radionuclide to form a complex and,
11 subsequently, the metal chelate conjugate formed can be
12 purified by, for example, size exclusion high pressure
13 liquid chromatography. A desirable hapten for the
14 ligand conjugate can be selected from the group
15 consisting of hormones, steroids, enzymes, and proteins.

16 The most commercially useful embodiments of the
17 invention are chelate conjugates having general formula I
18 wherein (1) n is an integer from 1 to 5, (2) X' is a
19 member selected from the group consisting of $-NHQ$,
20 $-NCS-Q$, $-NHCOCH_2-Q$, $-OCH_2COOQ$, and $-COO-Q$ with Q being a
21 hapten selected from the group consisting of hormones,
22 steroids, enzymes, and proteins, and (3) M is a metal ion
23 being a member selected from the group of elements
24 consisting of Bi, Pb, Y, Cd, Hg, Ac, Th, Sr, and
25 Lanthanides. These chelates conjugates can deliver
26 radioactive metal ions such as Pb^{212} , Bi^{212} , Y^{90} , Th^{224} ,
27 and Sr^{90} to specific cellular disorders.

28 The preferred embodiment of the invention uses a
29 chelate conjugate binding Pb^{212} . Pb^{212} is a very
30 desirable pharmaceutical compound for delivering both
31 beta and alpha radiation to a selected site for treatment
32 of the cellular disorders. The delivery is made through

1 the Pb²¹² ion which converts with a 10 1/2 hour half-life
2 into Bi²¹². Bi²¹² and daughters deliver one alpha
3 particle per Pb²¹² nucleus. The desirable result of this
4 chelate conjugate is that the Pb²¹² half-life is
5 sufficient to allow site selection from the body fluid by
6 the hapten before the alpha particle is emitted.

7 The invention includes a process for treating
8 cellular disorders. This process uses the chelate
9 conjugate with a hapten having a selective binding site
10 at the cellular disorder. For example, Q can be a
11 monoclonal antibody wherein the antibody is directed and
12 created against an epitope found specifically on the
13 tumor cells. Thus, when Pb²¹² is transported to the
14 antigen site and, subsequently, decays in secular
15 equilibrium to Bi²¹² and its daughters, a beta
16 irradiation is produced from the lead disintegration. A
17 beta radiation is produced by the bismuth daughters.
18 This beta radiation is similar to the beta radiation from
19 Y⁹⁰, but in addition, each disintegration of bismuth also
20 produces an alpha particle. In this manner, a
21 radiotherapy is provided with a radiation dose from both
22 an alpha and a beta particle. If desired, only Bi²¹² can
23 be introduced in those cases where the disorder to be
24 treated, such as with leukemic cells, can be easily
25 reached within the 1 hour half-life of Bi²¹². It is also
26 possible to use this method to treat cancers where the
27 cells are widely differentiated. This might be preferred
28 where only a long range beta emitter, such as Y⁹⁰, is
29 desired. In differing environments, in vivo, the Bi²¹²
30 is retained inside the chelate after the beta emission in
31 differing amounts. Most desirably, at least 95 percent
32 of Bi²¹² remains in the chelate. In an acidic medium,
33 such as the stomach, at least about 70 percent of the

1 Bi²¹² is retained. Retaining at least about 80 or 90
2 percent, Bi²¹² is also desirable depending on the medium.

3 The invention includes a process for diagnostic
4 testing. This process uses a chelate conjugate having
5 general formula I wherein M is a member selected from the
6 group consisting of Pb²⁰³, Tc^{99m}, In¹¹¹, Ga⁶⁷, Ga⁶⁸,
7 Sc⁴³, Sc⁴⁴, Fe⁵², Fe⁵⁴, Fe⁵⁶, Fe⁵⁷, Fe⁵⁸, and Co⁵⁵. The
8 usefulness of metal ions with both in vitro and in vivo
9 diagnostic procedures is disclosed in U.S. Patent Number
10 4,454,106, hereby incorporated by reference.

11 The most desirable embodiment of this diagnostic
12 process uses Pb²⁰³. Pb²⁰³ has a 52.1 hour half-life as a
13 gamma emitter. Pb²⁰³ has a unique property in that it
14 decays at a high percentage only by a single photon
15 emission. This gamma emission is preferred and dominant
16 over all other emissions. This single photon emission
17 makes Pb²⁰³ useful for single photon emission computed
18 spectroscopy [SPECT] which is a diagnostic tool. Thus,
19 when Pb²⁰³ is linked by use of the chelate to a hapten
20 which specifically localizes in a tumor, then that
21 particular localization can be three dimensionally mapped
22 for diagnostic purposes in vivo by single photon emission
23 tomography. Alternatively, the emission can be used in
24 vitro in radioimmunoassays.

25 EXAMPLE 1

26 The procedures and reagents described above for the
27 preferred embodiment of making the compounds are used for
28 this example.

1 The antibody specific for the IL-2 antigen is the
2 monoclonal antibody alpha-Tac. This antibody is labelled
3 with the chelate of compound XII of Figure 1 as follows.
4 The antibody is suspended in a buffered normal saline
5 solution having a pH of about 8.5. Solid ligand or
6 compound XII is added to the protein suspension. The
7 protein conjugate forms during reaction overnight and is
8 purified by dialysis against metal-free 0.05 molar
9 citrate/0.15 molar sodium chloride buffer at pH 5.5.
10 Before labelling with metal, the protein is dialyzed
11 against a solution comprising 0.02 molar
12 N-morpholinoethanesulfonic acid and 0.02 molar acetate at
13 pH 5.9.

14 The protein in solution is labelled with Y90 by
15 reacting with an acetate solution of the isotope followed
16 by passage through a TSK 3000 size exclusion column. This
17 is a high pressure liquid chromatography procedure. The
18 compound is mixed with a pharmaceutical excipient and is
19 used in mammals in a therapeutic amount to treat adult
20 T-cell leukemia in mammals. T-cell leukemia displays the
21 characteristics of having an extraordinarily large
22 amounts of IL-2 receptors located on the tumor cells. The
23 antibody localizes specifically to these tumor cells to
24 deliver its radiation.

25 EXAMPLES 2 and 3

26 The procedures and reagents described above for the
27 preferred embodiment of making the compounds are used for
28 these examples. The only difference between Example 1
29 and Examples 2 and 3 are the use of the antibody B72.3
30 which binds specifically to a glycoprotein on LS-174T
31 cells. This glycoprotein is also in humans which have
32 colon cancer. The model system of this example is an

1 athymic mouse which has been implanted with LS-174T cells
2 to develop a tumor on the flank of the animal in the
3 location of where the cells were implanted. The
4 diagnostic method used to visualize the growing tumor
5 involves the following components. The chelate of
6 compound 12 is first coupled to gadolinium or Pb203 by
7 mixture of the chelate solution at pH 4 to 5 with
8 gadolinium or Pb203 nitrate. This material can be then
9 linked directly to the antibody by mixture to react with
10 the protein and purified according to the method of the
11 previous example.

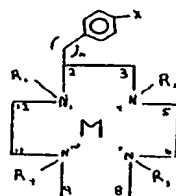
12 In Example 2, the gadolinium chelate ligand-protein
13 conjugate is injected or introduced into body fluids of a
14 mammal. The gadolinium then localizes along with the
15 antibody to the tumor and conventional resonance magnetic
16 imaging techniques are used to visualize the tumor.

17 In Example 13, the Pb203 is used, the metal labelled
18 protein conjugate is similarly introduced into the
19 mammal, but gamma camera or SPECT imaging is used to
20 visualize the tumor.

WHAT IS CLAIMED IS

1. A chelate comprising:

a general formula I:



wherein R_1-4 is $-\text{CH}_2\text{COOH}$;

wherein n is an integer from 1 to 5;

X is a member selected from the group consisting of

$-\text{NO}_2$,

$-\text{NH}_2$,

$-\text{NCS}$,

$\text{NHCOCH}_2\text{-Z}$ with Z being a member selected from the group consisting of Br and I

$-\text{OCH}_2\text{COOH}$;

$-\text{COOH}$;

and M is a metal ion being a member selected from the group of elements consisting of Bi , Pb , Y , Cd , Hg , Al , Th , Sr , and Lanthanides.

2. The chelate of claim 1 wherein n is 1 to 2 and X is a member selected from the group consisting of $-\text{NO}_2$, $-\text{NCS}$, and $-\text{NHCOCH}_2\text{-Z}$ with Z being a member selected from the group consisting of Br and I.

3. The chelate of claim 2 wherein M is a member selected from the group consisting of Bi, Pb, Y, Th, Sr, Gd, Eu, and Tb.

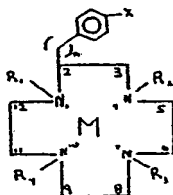
4. The chelate of claim 1 wherein n is 2, X is $-\text{NCS}$, and M is a member selected from the group consisting of Pb^{212} , Pb^{203} , Bi^{212} , Y^{90} , Th^{224} , and Sr^{90} .

5. The chelate of claim 4 wherein M is a member selected from the group consisting of Pb^{212} and Bi^{212} .

6. The chelate of claim 2 wherein n is 2, X is NCS , and M is a member selected from the group consisting of Eu and Tb.

7. A chelate comprising:

a general formula I:



wherein $\text{R}_1\text{-}_4$ is $-\text{CH}_2\text{COOH}$;

wherein n is an integer from 2 to 5;

X is a member selected from the group consisting of

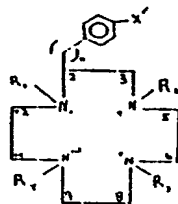
- 1 -NO₂,
 2 -NH₂,
 3 -NCS,
 4 -NHCOCH₂ - Z with Z being a member selected from
 5 the group consisting of Br and I
 6 -COOH;
 7 -OCH₂COOH;
 8 and M is Cu.

9 8. The chelate of claim 7 wherein n is 2 and X is a
 10 member selected from the group consisting of -NO₂, -NCS
 11 and -NCOCH₂-Z with Z being a member selected from the
 12 group consisting of Br and I.

13 9. The chelate of claim 8 wherein X is -NCS.

14 10. A ligand-hapten conjugate comprising:

15 a general formula II:



- 16 wherein R₁₋₄ is -CH₂COOH;
 17 n is an integer from 1 to 5;
 18 X' is a member selected from the group
 19 consisting of -NH-Q,
 20 -N-C-Q,
 21 | ||
 22 H S
 23 H
 24 |
 25 -NCOCH₂-Q,
 26 -OCH₂COOQ, and
 O
 ||
 -C-O-Q

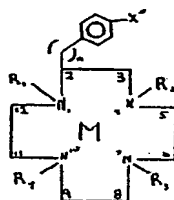
with Q being a hapten selected from the group of hormones, steroids, enzymes, and proteins.

11. The ligand conjugate of claim 10 wherein n is 2 and X' is $-\text{NH}-\overset{\text{O}}{\underset{\text{S}}{\text{C}}}-\text{Q}$.

12. The ligand conjugate of claim 11 wherein Q is a protein, said protein being a monoclonal antibody.

13. A chelate conjugate comprising:

a general formula I:



wherein R₁-4 is $-\text{CH}_2\text{COOH}$;

n is an integer from 1 to 5;

X' is a member selected from the group consisting of $-\text{NH}-\text{Q}$,

$-\text{NHCS}-\text{Q}$,

$-\text{NHCOCH}_2-\text{Q}$,

$-\text{OCH}_2\text{COOQ}$, and

$-\overset{\text{O}}{\underset{\text{O}}{\text{C}}}-\text{O}-\text{Q}$

with Q being a hapten selected from the group consisting of hormones, steroids, enzymes, and proteins;

M is a metal ion being a member selected from the group of elements consisting of Bi, Pb, Y, Cd, Hg, Ac, Th, Sr, and Lanthanides.

1 14. The chelate conjugate of claim 13 wherein n is 1
2 to 2 and X' is -NH-C-Q.
3



4 15. The chelate conjugate of claim 14 wherein M is a
5 member selected from the group of Bi, Pb, Y, Th, Sr, Gd,
6 Eu, and Tb.

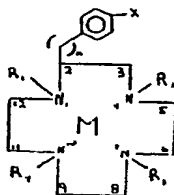
7 16. The chelate conjugate of claim 15 wherein n is 2
8 and M is a member selected from the group consisting of
9 Pb²¹², Pb²⁰³, Bi²¹², Y⁹⁰, Th²²⁴, and Sr⁹⁰.

10 17. The chelate conjugate of claim 16 wherein M is a
11 member selected from the group consisting of Pb²¹² and
12 Bi²¹².

13 18. The chelate conjugate of claim 15 wherein n is 2
14 and M is a member selected from the group consisting of
15 Eu and Tb.

16 19. A process for treating cellular disorders
17 comprising:

18 introducing into body fluid a solution of metal
19 chelate of a general formula I:



20 wherein R₁₋₄ is -CH₂COOH;
21 n is an integer from 1 to 5;
22 X is a member selected from the group consisting
23 of -NH-Q,

1 -NH-C-Q,

2 ||
 S

3 -NHCOCH₂-Q,

4 -OCH₂COOQ,

5 O
6 ||
 -C-O-Q

7 with Q being a hapten selected from the group consisting
8 of hormones, steroids, enzymes, and proteins;

9 M is a metal ion being a member selected from the group
10 of elements consisting of Bi, Pb, Y, and Gd.

11 20. The process of claim 19 wherein n is 2, X is
12 -NH-C-Q, and M is Gd.

13 ||
 S

14 21. The process of claim 19 wherein M is Pb²¹², said
15 Pb²¹² converts to Bi²¹² and daughters thereof, said Bi²¹²
16 emits an alpha particle to treat said cellular disorders.

17 22. The process of claim 21 wherein n is 2 and X is
18 -NH-C-Q.

19 ||
 S

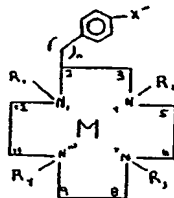
20 23. The process of claim 22 wherein at least 70
21 percent of said Bi²¹² remains in said chelate.

22 24. The process of claim 23 wherein at least 80
23 percent of said Bi²¹² remains in said chelate.

24 25. The process of claim 24 wherein at least 90
25 percent of said Bi²¹² remains in said chelate.

26. The process of claim 25 wherein at least 95 percent of said Bi212 remains in said chelate.

27. A process for diagnostic testing comprising:
introducing into a test medium a solution of metal chelate of a general formula I:



wherein R₁₋₄ is -CH₂COOH;

n is an integer from 1 to 5;

X' is a member selected from the group consisting of -NH-Q,

-N-C-Q,
||
S

-NHCOCH₂-Q,

-OCH₂COOQ, and

-C(=O)-Q with Q being a hapten selected from the group consisting of hormones, steroids, enzymes, and proteins;

M is a metal ion being a member selected from the group consisting of Pb203, Tc99m, In111, Ga67, Ga68, Sc43, Sc44, Fe52, Fe54, Fe56, Fe57, Fe58, and Co55.

28. The process of claim 27 wherein n is 2, X is -NCS-Q, and Q is a monoclonal antibody.

29. The process of claim 28 wherein M is Pb203.

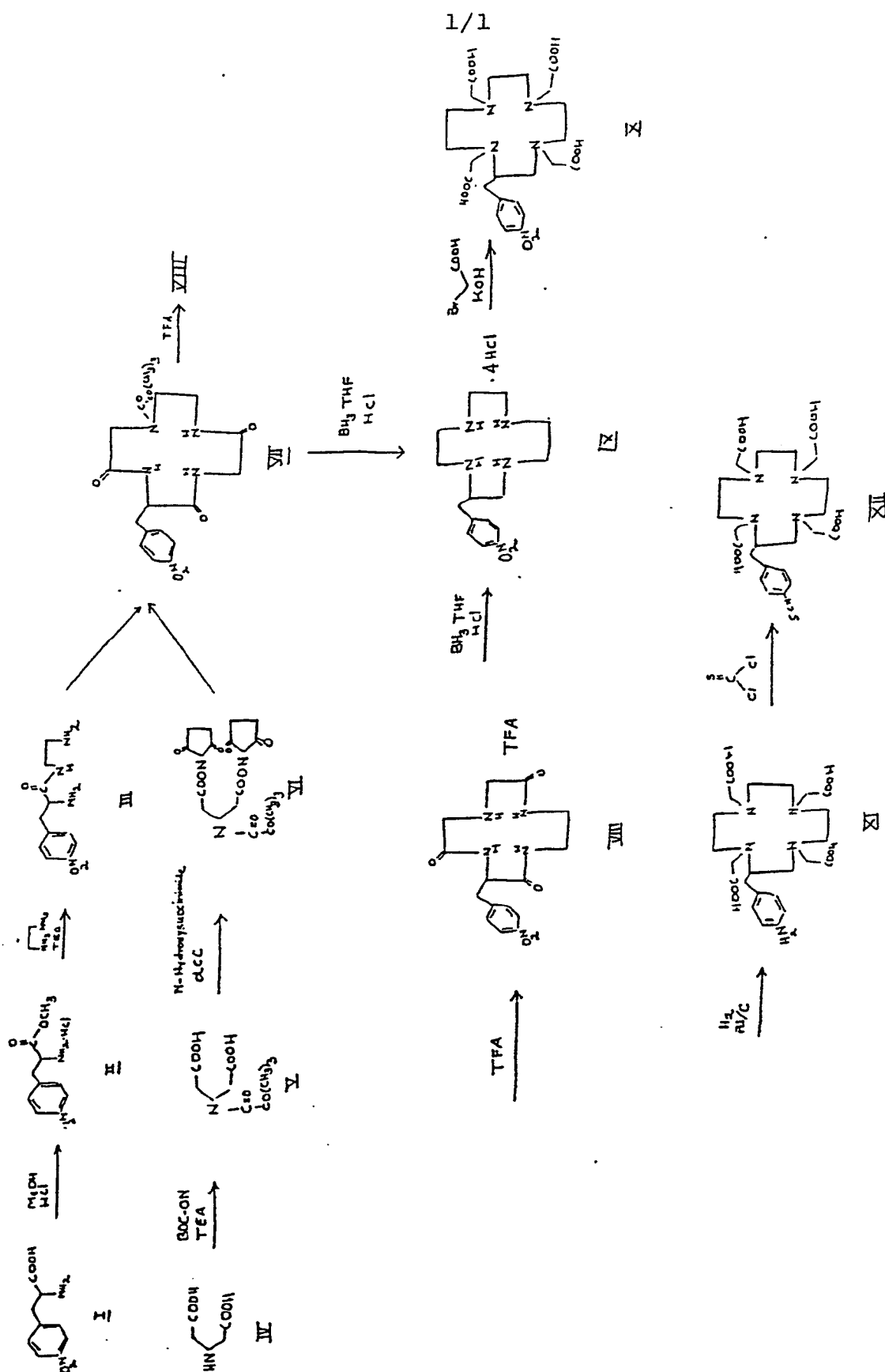



FIGURE 1

INTERNATIONAL SEARCH REPORT

International Application No. PCT/US89/02201

| | | |
|--|--|--|
| I. CLASSIFICATION OF SUBJECT MATTER (If several classification symbols apply, indicate all) * | | |
| According to International Patent Classification (IPC) or to both National Classification and IPC IPC(4): C07D 257/00; C07E 9/94, 7/24, 5/06, 3/08, 3/10, 5/00, 3/12, 3/00 U.S.C1.: 540/465, 474; 534/15 | | |
| II. FIELDS SEARCHED | | |
| Minimum Documentation Searched ⁷ | | |
| Classification System | Classification Symbols | |
| U.S. | 540/465, 474; 534/15 | |
| Documentation Searched other than Minimum Documentation to the extent that such documents are included in the fields searched ⁸ | | |
| Chemical Abstracts - manual - Vo. 1 (1907)- Vo. 109 (to date) CA on-line (computer)-1967-1989 | | |
| III. DOCUMENTS CONSIDERED TO BE RELEVANT ⁹ | | |
| Category * | Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹² | Relevant to Claim No. ¹³ |
| A | U.S., A, 3,936,445 (PFITZNER. ET AL) published 3 February 1976. See entire document. | 1-9 |
| <div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <p>* Special categories of cited documents: ¹⁰</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> </div> <div style="width: 45%;"> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&" document member of the same patent family</p> </div> </div> | | |
| IV. CERTIFICATION | | |
| Date of the Actual Completion of the International Search | | Date of Mailing of this International Search Report |
| 23 AUGUST 1989 | | 01 SEP 1989 |
| International Searching Authority | | Signature of Authorized Officer |
| ISA/US | |  ROBERT C. WHITTENBAUGH |

Form PCT/ISA/210 (second sheet) (Rev.11-87)

FURTHER INFORMATION

CONTINUED FROM THE SECOND SHEET

V. ☐ OBSERVATIONS WHERE CERTAIN CLAIMS WERE FOUND UNSEARCHABLE¹

This international search report has not been established in respect of certain claims under Article 17(2) (a) for the following reasons:

1. ☐ Claim numbers _____, because they relate to subject matter ¹² not required to be searched by this Authority, namely:

2. ☐ Claim numbers _____, because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out ¹³, specifically:

3. ☐ Claim numbers _____, because they are dependent claims not drafted in accordance with the second and third sentences of PCT Rule 6.4(a).

VI. ☒ OBSERVATIONS WHERE UNITY OF INVENTION IS LACKING²

This International Searching Authority found multiple inventions in this international application as follows:

Lack of unity of invention has been found to exist in view of the following independent and distinct inventions. The distinct inventions are as follows:

(I) Chelates in which M is a heavy or light metal embraced by claims 1-9 and classified in 540-465. (See Attachment sheet 1)

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims of the international application.

2. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims of the international application for which fees were paid, specifically claims:

3. ☒ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claim numbers: 1-9

4. ☐ As all searchable claims could be searched without effort justifying an additional fee, the International Searching Authority did not invite payment of any additional fee.

Remark on Protest

☐ The additional search fees were accompanied by applicant's protest.

☐ No protest accompanied the payment of additional search fees.

PCT/US89/02201
Attachment sheet 1.

VI. OBSERVATIONS WHERE UNITY OF INVENTION IS LACKING
(CONTINUED)

- (II) Ligand-hapten conjugates and methods for treating cellular disorders embraced by claims 10-26 and classified in 435-188, 530-404, and 436-543.
- (III) Methods of diagnostic testing embraced by claims 27-29 and classified in 424-2.

The claims of these 3 groups are directed to different inventions which are not so linked as to form a single general inventive concept. Each group is drawn to an independent and distinct invention.

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